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2. Carbonyl is a general term for anything with the formula -- -- whether aldehyde or ketone.
3. A carboxyl does not have the same properties as a carbonyl plus a hydroxyl. The H in a carboxyl
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6 Properties of Covalent Compounds
Covalent bonds between atoms are strong, but attraction forces between molecules are weak. The weak attraction

Forces Are Known As Van Der Waals Forces. The Forces Vary In Strength But Are Weaker Than The Bonds In A Molecule Or Ions In An Ionic Compound Non-polar Molecules Jan 8th, 2024.

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Equations 36 190 35 194xx 12 45 68 Xx 1. Take The Number On The Left To Zero. 2. Do The Same

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