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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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2 H₂O + 2 K₂SO₄ (8) + BaCl₂ + KCl + BaSO₄ CsH₁₂ (D₂ + CO₂ + H₂O Ca
(OH)₂ (N) + NHCl (D²D^{3/4}D^{1/2}N̄ D¹ N̄ D^o N̄ N̄ D²D^{3/4}N̄) + CaCl NHOHp) VO₅ (4) +
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Chapter 7 Chemical Reactions Section 7.1 Describing Reactions

Measure Amounts Of A Substance Because Chemical Reactions Often Involve Large Numbers Of Small Particles. 10. Circle The Letter Of The Correct Answer. If One

Carbon Atom Has An Atomic Mass Of 12.0 Amu And One Oxygen Atom Has An Atomic Mass Of 16.0 Amu, What Is The Molar Mass Of Carbon Dioxide? A.28.0 Amu B.44.0 Amu C.28.0 Gd.44.0 G 11. Jan 3th, 2023

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