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And Try.  $2\text{Ba} + \text{O}_2 \rightarrow 2\text{BaO}$  35. The Number Of Atoms  
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Of The Equation Once It Has. Balanced Equations. Six  
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Answers Balancing Chemical Equations Worksheet 1 -

ANSWERS 1.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  2.  $2\text{Na} + \text{Cl}_2 \rightarrow$

$2\text{NaCl}$  3.  $\text{N}_2\text{O}_4 \rightarrow 2\text{NO}_2$  4.  $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$  5.  $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$  6.  $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$  7.  $2\text{Li} +$

$\text{F}_2 \rightarrow 2\text{LiF}$  8.  $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$  9.  $2\text{NH}_3 \rightarrow 3\text{N}_2 +$

$2\text{H}_2$  10.  $2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$  ... Feb 10th, 2024

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The Equations Below: 1)  $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  2)  $2$

$\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  3)  $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$

4)  $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$  5)  $1\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O}$

$+ 1\text{PbCl}_2$  6)  $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$  7)  $1\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O}$  Apr 9th, 2024

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