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12): Sections 5.1 (p298 - 305); P294 #3, 6, 7; P302 #10 Answers: 1. 7.35 KJ 2.  
3.1x10<sup>2</sup> KJ 3. 135 J 4. 43.5°C 5. C = 0.235 J/g • °C 6. 25.91 J • Mol<sup>-1</sup> • °C<sup>-1</sup> 7. A) 3.8  
X 10<sup>3</sup> J B) 53 KJ/mol 8. 23.41°C. Title: SCH4U-Calorimetry Worksheet Author ... Mar  
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Of The Heat Flow Into Or Out Of A System For Chemical And Physical Processes.  
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Is The Heat Of Combustion Of Benzoic Acid (given As -6318 Cal G), M Is The Mass Of The Benzoic Acid Sample (0.9799 G), E 3 Is The Heat Of Combustion Of The Wire (calculable From The Heat Of Combustion Of The Wire, -2.3 Cal Cm And The Length Of The Wire, 15.1 Cm), And T Is The Previously Calculated Temperature Rise, 2.77252 C . Combining Mar 2th, 2024 Experiment 9: CALORIMETRY  $\Delta T = \text{Change In Temperature Of The Solution} = T_{\text{Final}} - T_{\text{Initial}}$  In This Experiment Mass Of The Solution Will Be Calculated Using The Volume Of The Solution And Its Density (assumed To Be 0.997 G ML<sup>-1</sup>) Heat Of Neutralization Is The Same Value As Q Soln But With The Opposite Sign. The Solution Is Absorbing Heat So Q Soln Is Positive. Mar 3th, 2024.

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Waterman (1971), Discussed Below, In Which The Relationship Between Oxygen Depletion In A Hooded Exhaust Was Directly Proportional To The Rate Of Oxygen Consumption. Apr 2th, 2024.

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Compound And The Molar Heat Of Combustion So By Multiplying Them Together We

Can Find The Heat Generated.  $(0.875 \text{ Mol})(1350 \text{ KJ/mol}) = 1180 \text{ KJ} = 118000 \text{ J}$  Plug This Value Into Our Equation And Then Rearrange To Find The Change In The Jan 3th, 2024

Calorimetry: Measuring The Energy In Foods

1. Using The Graduated Cylinder, Obtain 50 ML Of Water And Carefully Pour It Into The Soda Can.
2. Determine The Mass Of Water And The Can. Record The Mass Of Water In The Data Table (hint: Density Of Water = 1 G/mL).
3. Hold The Paper Clip Horizontally And Bend The Outer End Upwards Until Mar 3th, 2024.

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