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(c) (i) Write A Chemical Equation Representing The
Lattice Energy Of AgBr. (ii) Use The Following Data To
Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = $+731\text{kJmol}^{-1}$ Electron
Affinity Of Bromine = 325kJmol^{-1} Enthalpy Change Of
Atomisation Of Silver = $+285\text{kJmol}^{-1}$ Enthalpy Change
Of Atomisation Of Bromine = $+112\text{kJmol}^{-1}$... Mar 4th,
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