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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To



Calculate A Value For The Lattice Energy Of AgBr(s).  
First Ionisation Energy Of Silver =  $+731\text{kJmol}^{-1}$  Electron  
Affinity Of Bromine =  $325\text{kJmol}^{-1}$  Enthalpy Change Of  
Atomisation Of Silver =  $+285\text{kJmol}^{-1}$  Enthalpy Change  
Of Atomisation Of Bromine =  $+112\text{kJmol}^{-1}$  ... Feb 25th,  
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