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Answer On Question #60099, Chemistry / Inorganic Chemistry 1. 1.00g Of Carbon Is Combusted In A Limited Supply Of Pure Oxygen. 0.50g Of The Carbon Combusts To Form CO₂ And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture Of CO₂ And CO Is Passed Through Excess Na Mar 5th, 2024

Answer On Question #58182 Chemistry Organic Chemistry ...

2) Aniline and Acetanilide stand in a one-to-one molar relationship, so 0.0322 mol of aniline produces 0.0322 mol of acetanilide. The amount of acetic anhydride does not play a role, since it is in excess. Then, $n(\text{aniline}) = n(\text{Acetanilide}) = 0.0322$ mol. We find the molar mass of acetanilide: $M(\text{Acetanilide}) = 12 + 8 + 1 + 9 + 14 + 1 + 16 + 1 + 135 = 187$ g/mol. Apr 4th, 2024

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21. An example of a physical change is . 1) Burning; 2) Decaying; 3) Melting; 4) Rusting; 5) Oxidation. ANS: 3 . 22. In a physical change of matter, 1) a chemical change takes place; 2) chemical properties change. Jan 4th, 2024

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