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Hydrates Practice - AP Chemistry - Homepage

Problem #5: Given That The Molar Mass Of $\text{Na}_2\text{SO}_4 \cdot \text{NH}_2\text{O}$ Is 322.1 G/mol, Calculate The Value Of N. Solution: 1) The Molar Mass Of Anhydrous Na_2SO_4 Is: 142.041 G/mol 2) The Mass Of Water In One Mole Of The Feb 7th, 2024

CHEM 10100 - Inquiries Into The Nature Of Matter

In The First Day Of Lab You Will Be Required To Have: • Chemistry 101 "Inquiries into The Nature Of Matter" Laboratory Manual By D. McGregor, P. Mills, G. Smeureanu
Available For Purchase Jan 7th, 2024

Naming Acids And Hydrates Worksheet

Naming Bases Most Strong Bases Contain Hydroxide, A Polyatomic Ion. Therefore, Strong Bases Are Named Following The Rules For Naming Ionic Compounds. For Example, NaOH Is Sodium Hydroxide, KOH Is Potassium Hydroxide, And $\text{Ca}(\text{OH})_2$ Is Calcium Hydroxide. Weak Bases Made Of Ionic Compounds Are Available Apr 1st, 2024

Hydrates 2 - Worksheet