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Worksheet 16 - Equilibrium  
Chemical Equilibrium  
Worksheet 16 - Equilibrium  
Chemical Equilibrium Is The State Where The Concentrations Of All Reactants And Products Remain Constant With Time. Consider The Following Reaction:  
$$\text{H}_2\text{O} + \text{CO} \rightleftharpoons \text{H}_2 + \text{CO}_2$$
  
Suppose You Were To Start The Reaction With Some Amount Of Each Reactant (and No H Apr 5th, 2024  
Static Equilibrium For Forces  
Static Equilibrium And G GGG ...
$$F_{\text{Pivot}} = (m_B + m_1 + m_2)g$$
  
$$F_{\text{Pivot}} - m_B g - N_{B,1} - N_{B,2} = 0$$
  
Worked Example: Solution  
Pivot Force: Lever Law:  $F_{\text{Pivot}} = (m$

$B + m_1 + m_2)g = (2.0 \text{ Kg} + 0.3 \text{ kg} + 0.6 \text{ Kg})(9.8 \text{ M} \cdot \text{s}^{-2})$   
 $= 28.4 \text{ N}$   
 $D_1 M_1 = d_2 M_2$   
 $D_2 = d_1 m_1 / M_2 = (0.4 \text{ M})(0.3 \text{ Kg} / 0.6 \text{ Kg}) = 0.2 \text{ M}$   
Generalized Lever Law, , 1  
11 22, 2,  $\perp \perp = + = +$  FF F FF F & & GG G GGG Mar 3th, 2024.

Equilibrium Process Practice Exam Equilibrium Name (last ...A)  $K_{eq}$  1 D)  $K_{eq}$  Cannot Be Determined. 6  
Concentration And Solubility Of Gas The Solubility Of CO<sub>2</sub> Gas In Water Is 0.240 G Per 100 ML At A Pressure Of 1.00 Atm And 10.0°C. May 2th, 2024  
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