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ANSWER KEY H = H (Products) - H (Reactants) 0 0 0

FfG = 3.14 X10 J/mol B) If Requires 193 J To Vaporize 1.00 G Of Liquid Bromine At 25 C And 1.00 Atm. Calculate H And S At 25 C For This Reaction. H = X = 3.084 X10 J/mol 193 J 159.8 G 4 1 G 1 Mol H - G (30840 - 3140) J/mol Feb 25th, 2024Excess Air = Excess OxygenO . For 25% Excess Air In The Example, We Feb 22th, 2024KEY CONCEPT The Masses Of Reactants And Products Are Equal.Chapter 3: Chemical Reactions 81 DB Chemical Equations Must Be Balanced. Remember, Chemical Reactions Follow The Law Of Conservation Of Mass. Chemical Equations Show This Conservation, Or Equality, In Terms Of Atoms. The Same Number Of Atoms Of Each Element Must Appear On Both Sides Of A Che Feb 22th, 2024.

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