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CHEM 1413 Homework Questions TEXTBOOK HOMEWORK ... 5. A 96 g Sample Of Methanol Gas, CH_3OH [$M = 32$] Is Initially At 200 °C And 700 Torr. The Sample Is Cooled At

Constant Pressure To 120 OC. During This Process, The Volume Decreases From 126. L To 105 L. The Constant Pressure Molar Heat Capacity Of Methanol Is 52 J/ Feb 6th, 2024
www.tutor-homework.com (for Tutoring, Homework Help, Or ...Decomposes In The Gas Phase By The Reaction $\text{SO}_2 + \text{Cl}_2 (\text{g}) \rightarrow \text{SO}_2 (\text{g}) + \text{Cl}_2 (\text{g})$ The Reaction Is First Order In SO_2 And The Rate Constant Is $3.0 \times 10^{-6} \text{ s}^{-1}$ at 600 K. A Vessel Is Charged With 3.3 Atm Of At 600 K. The Partial Pressure Of SO_2 At Is ____ Atm. Student Response Mar 4th, 2024
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