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Potassium Chlorate (KClO_3) Is Heated, It Decomposes To Form Gases KCl And Oxygen (O_2 .) $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$ C. Jan 20th, 2024
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33. $\text{KClO}_4 \rightarrow 2 \text{O}_2 + \text{KCl}$ Actual yield = 1.45g

The yield = $5.95 \text{ g KClO}_4 \times \frac{1 \text{ mol}}{122.55 \text{ g}} \times 2 \text{ O}_2 \times \frac{1 \text{ KClO}_4}{4 \text{ O}_2} \times 32.00 \text{ g/mol} = 3.11 \text{ g O}_2$
% Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$
 $34.4 \text{ FeCO}_3 + \text{O}_2 \rightarrow 2 \text{ Fe}_2\text{O}_3 + 4 \text{ CO}_2$
Percent purity = 62.8%
pure FeCO_3 Apr 17th, 2024
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In A Camping Stove? 2. How Many Moles Of O_2 Molecules Should Be Supplied To Burn 1 Mol Of CH_4 ...
Was At One Time Considered For Use As A Rocket Fuel. ... How Many Formula Units Of Calcium Oxide (CaO)
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Answers 44. What Number Of Moles Of O₂ Is Needed
To Produce 14.2 Grams Of P₄O₁₀ From P? (Molar
Mass P₄O₁₀ = 284) (A) 0.0500 Mole (B) 0.0625 Mole
(C) 0.125 Mole Apr 23th, 2024.

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