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In Liquid State. This Is Due To. A) High Boiling Point B)
Low Boiling Point C) Freezing Point Is Zero D) Hydrogen

Bond 11. The Hybridization In SO_4^{2-} ion Is A) sp^3 B) $\text{sp}^3 \text{d}^3$ C) $\text{sp}^3 \text{d}$ D) $\text{sp}^3 \text{d}^2$ 12. Number Of Spherical Nodes In 2s Orbital Is A) 1 B) 2 C) 3 D) 4

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MARKS 1 C) $\text{CO} + \text{H}_2$ 1 2 B) 2 1 4 3 1 3 A) 1 Sum Of

Molar Heat Of Fusion And Vaporisation 4 A) 4,6 1 5

B) NH_3 1 6 A) B) +6 1 7 B) D) -9E 1 8 C) +3 KJ 1 9 A) iv <

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