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Name Each Of The Compounds: (a)  $\text{CA}$  ( $\text{H}_2\text{PO}_4$ ) 2 (b)

FESO<sub>4</sub> (C) CaCO<sub>3</sub> (D) MgO (E) NaNO<sub>2</sub> (F) KI  
Response To A Calcium Dihydrogen Phosphate Answer B Isolated  
(II) Sulphate Response C Carbonate Football Ansonual  
Reply Of Magnesium Oxide And Sodium Nitrite Reply F  
Of Iodide Potassium ã, Contribut Feb 20th, 2024

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Write The Formula For Each Covalent Compound.  
Iodine Trichloride Disulfur Dibromide Arsenic Trioxide  
Xenon Hexafluoride Write The Formula For Each  
Covalent Compound. Boron Trichloride Carbon Dioxide  
Tetraphosphorus Decoxide Germanium Dichloride  
Write Two Covalent Jan 2th, 2024

### **Nomenclature Ultimate Review 2017-Answers**

Part C — Simple Binary Covalent Compounds I. Provide  
Names For The Following Covalent Compounds: 1. NF<sub>3</sub>  
Nitrogen Trifluoride 2. XeO<sub>3</sub> Xenon Trioxide 3. OF<sub>2</sub>  
Oxygen DiFluoride 4. N Mar 4th, 2024

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For Example:  $[\text{FeF}_6]^{3-}$ , Hexafluoroferrate (III). Note Here Also That The Latin Name For Iron Is Used To Avoid An Awkward Sounding Name. A List Of Such Transition Metals Follows In The Section On Naming Rules. Ligands Are Ions, Ionic Molecules, Or Neutral Molecules That Have Atoms Which Poss Jan 20th, 2024

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