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2. Q. List Three Subatomic Particles Of An Atom. Compare Them On The Basis Of Relative Mass And Charge In A Tabular Form. Ans: Particle Symbol Charge Mass Electron E---1.60×1019 C  $9.1\times10-31$  Kg Proton +p (H+)  $1.60\times10-1$  Apr 7th, 2024

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(c) Three Mole (d) Four Moles (4) Which Of The Following Weighs The Most? (a) One G-atom Of Nitrogen (b) One Mole Of Water (c) One Mole Of Sodium (d) One Molecule Of H2So4 (5) 5.0 Litre Of Jan 1th, 2024

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