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(1) Weigh 1.0 1.2 G (precision Of ± 0.0002 G) Of $\text{Cu}(\text{gly})_2 \cdot x\text{H}_2\text{O}$ With A Dry 100 cm^3 Beaker As The Container. (2) Dissolve It With 40 cm^3 Of Water And 8 cm^3 Of H_2SO_4 (1.0 $\text{mol} \cdot \text{dm}^{-3}$). (3) Transfer The Above Solution Quantitatively To A 100 cm^3 Volumetric Feb 3th, 2024

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