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2 Has A Bond Order Of 1 Adding An Electron Will Add To An Anti-bonding Orbital And Decrease The Bond Order To 0.5 Removing And Electron Will Remove From An Anti-bonding Orbital And Increase The Bond Order To 1.5 Therefore F 2+ Will Have The Highest Bond Order And Strongest Bond F 2-will Have The Weakest Bond Feb 11th, 2024

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C) 1,1-dibromo-2-chloro-2-fluoropropane D)1,1-dibromo-2,2 Dichlorofluoropropane 12 Ethyne Is A(n) A) Alkane B) Alkene C) Alcohol D) Aromatic E) Alkyne 13 How Many Alkyne Isomers Are Possible With The Molecular Formula C6H10? A) 3 Or Less B) 4 C) 5 D) 6 E) 7 Or More 14 Alkanes Have The General Formula A) CnH2n+2 B) CnH2n C) CnH2n-2 D) CnHn E ... Apr 5th, 2024

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