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Thermochemistry, Thermodynamics

Thermochemistry

Standard States The thermodynamic standard state of a substance is its most stable pure form under standard pressure (1 bar) and some specified temperature (298 K unless otherwise specified). For a pure substance in the liquid or solid phase, the standard state is the pure substance at 1 bar and the specified temperature. Feb 1th, 2024

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- (13 2. A Quantity Of Water Vapor At 1000C Is

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Releases 6.50×10^4 J Of Energy. What Mass Of ... Jan

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$\text{C}_2\text{H}_5\text{OH}(l) + 3 \text{ O}_2(g) \rightarrow 2 \text{ CO}_2(g) + 3 \text{ H}_2\text{O}(g)$ $\Delta H^\circ_{\text{rxn}} = -1366.8 \text{ kJ/mol}$ 31) A 100.0 mL Sample Of 0.300 M NaOH Is Mixed With A 100.0 mL Sample Of 0.305 M HNO_3 31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 °C And The Temperature Of The Resulting Solution Was Recorded As 37.0 °C, Determine The ΔH_{rxn} (in Units Of kJ/mol). Mar 9th, 2024

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