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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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2 H₂O + 2 K₂SO₄ (8) + BaCl₂ + 2 KCl + BaSO₄ + CsH₁₂ (D₂ + CO₂ + H₂O Ca
(OH)₂ (N) + NHCl (D₂D^{3/4}D^{1/2}N¹ N⁰ N¹ N¹ D²D^{3/4}N¹) + CaCl NHOHp) VO₅ (4) +
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