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Stopper The Calorimeter Tightly (no Gas Is Evolved In This Reaction) And Gently Swirl The Mixture. Record The Maximum Temperature Reached By The Mixture. Weigh The Vial With Any Residue. Find The Total Mass Of MgO Used In The Experiment By Difference. Repeat The Procedure. 1 L Mg 1 G Mg G Mg Q1 H1 = Calculation Of H 2 Calculate Q For The Second Experiment In The Same Manner As For The First ... Mar 4th, 2024

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The Experimental Evidence Collected Was Recorded Below: Mass Of Calorimeter 25.45 G Mass Of Calorimeter And Water 175.45 G Mass Of Ammonium Nitrate 1.68 G Initial Temper Jan 7th, 2024

Chemistry Unit 7 Worksheet 2: Enthalpy Of Reactions

3. (1 Mark Each) When 10.0 G Of Ammonium Nitrate Is Dissolved In 100 Cm³ Of Water, The Temperature Of The Solution Decreases From 19.0°C To 10.5°C. A Compare The Total Enthalpy Of The Reactants With That Of The Products And State Which Is Greate Jan 16th, 2024

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Diagram 1 Diagram 2 Diagram 3 - Drtayeb.files.wordpress.com

Number Of Sides Name Of Polygon Total Number Of Diagonals 3 Triangle 0 4 Quadrilateral 2 5 5 6 Hexagon 9 7 Heptagon 14 8 [3] (b) Write Down The Total Number Of Diagonals In (i) A Decagon (a 10-sided Polygon), Answer(b)(i) [1] (ii) A 12-sided Polygon. Answer(b)(ii) [1] Feb 16th, 2024

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