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Enthalpy The Heat Content A Substance Has At A Given Temperature And Pressure Can't Be Measured Directly Because There Is No Set Starting Point The Reactants Start With A Heat Content The Products End Up With A Heat Content So We Can Measure How Much Enthalpy Changes Feb 5th, 2024

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CHAPTER 6 THERMOCHEMISTRY

162 CHAPTER 6: THERMOCHEMISTRY To Convert The Answer To Joules, We Write: 101.3 J 0.18 L Atm 1L Atm $=-\cdot \times = \cdot$ W -18 J 6.17 An Expansion Implies An Increase In Volume, Therefore W Must Be -325 J (see The Defining Equation For Pressure-volume Work.) If The System Absorbs Heat, Q Must Be +127 J. The Change In Energy (internal Feb 5th, 2024

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6.6 Enthalpy: The Heat Evolved In A Chemical Reaction At Constant Pressure 210 6.7 Constant-Pressure Calorimetry: Measuring Δ RH 214 6.8 Relationships Involving Δ RH 216 6.9 Determining Enthalpies Of Reaction From Standard Enthalpies Of Formation 218 6.10 Energy Use Feb 4th, 2024

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