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$\text{C}_2\text{H}_5\text{OH}(l) + 3 \text{O}_2(g) \rightarrow 2 \text{CO}_2(g) + 3 \text{H}_2\text{O}(g)$  Cot 31) A 100.0 ml. Sample Of 0.300 M NaOH Is Mixed With A 100.0 ml. Sample Of 0.305 M  $\text{HNO}_3$  31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 °C And The Temperature Of The Resulting Solution Was Recorded As 37.0 °C, Determine The  $\Delta H_{\text{rxn}}$  (in Units Of kJ/mol). Feb 20th, 2024

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