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C2H5OH(l) + 3 2 CO2(g) + 3 H2O(g) Cot 31) A 100.0 Ml. Sample Of 0.300 M NaOH Is Mixed With A 100.0 Ml. Sample Of 0.305 M HNO3 31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 Oc And The Temperature Of The Resulting Solution Was Recorded As 37.0 Cc, Determine The AHrxn (in Units Of KJ/mol). Mar 23th, 2024

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### **THERMOCHEMISTRY**

SECTION 17.2 MEASURING AND EXPRESSING ENTHALPY CHANGES (pages 511-517) This Section Explains How To Construct Equations And Perform Calculations That Show Enthalpy Changes For Chemical And Physical Processes. ~ Calorimetry (pages 511-513) 1. The Property That Is Useful For Keeping Track Of heat Transfers In Chemical Jan 21th, 2024

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